

XPS characterization of functionalized epoxid microparticles

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Microparticles made of cross-linked polymers play an important role in a variety of applications such as medicine, chromatography and organic synthesis. One important parameter for the performance of microparticles in these applications is the availability of functional (polymeric) groups on the particle's surface. Two different syntheses are here presented: on the one hand boronic acid functionalized microparticles are prepared for the first time via mild epoxide ring opening based on porous cross-linked polymeric microparticles [1] and on the other hand porous poly(glycidyl methacrylate) (PGMA) microparticles are functionalized via hetero Diels-Alder (HDA) chemistry using electron deficient thiocarbonyl thio compounds [2].

In this study, X-ray photoelectron spectroscopy (XPS) revealed detailed information about the chemical composition of the modified microparticle surfaces and contribute significantly to the successful verification of key synthetic steps. Further employed scanning electron microscopy (SEM) corroborates the successful modification of the microparticles.

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Literatur

1. A. P. Vogt, V. Trouillet, A. M. Greiner, M. Kaupp, U. Geckle, L. Barner, T. Hofe, C. Barner-Kowollik, *Macromol. Rapid Commun.* **2012**, *33*, 1108–1113.
2. M. Kaupp, A. P. Vogt, J. C. Natterodt, V. Trouillet, U. Geckle, T. Hofe, L. Barner, C. Barner-Kowollik, *Polym. Chem.* **2012**, DOI: 10.1039/c2py20369c.