## INFLUENCE OF CHEMISORPTION PRODUCTS OF CARBON DIOXIDE ON RADIOLYSIS OF TRITIUM BREEDING CERAMIC

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Lithium orthosilicate pebbles (90mol%  $Li_4SiO_4$  and 10mol%  $Li_2SiO_3$ ) are internationally approved tritium breeding ceramic for fusion reactors [1]. On the surface of the pebbles tritium accumulates in  $T^+$  form, due to the interaction with radiation defects and products of radiolysis [2]. This effect could be explained by the fact that a radiation unstable  $Li_2CO_3$  phase forms in chemisorbing process of carbon dioxide from air atmosphere [3,4,5]. The aim of the investigation was to estimate the influence of chemisorption products of  $CO_2$  on the radiolysis of  $Li_4SiO_4$  containing ceramic.

Lithium orthosilicate powder with relevant composition was selected as investigation material because of the high specific surface area and small grain sizes ( $S_{\text{spec.}}=17\pm2~\text{m}^2\cdot\text{g}^{-1}$ ,  $\varnothing_{\text{powd.}}=350\text{-}550~\text{nm}$ ). The powder was thermally treated ( $T_{\text{max}}=300^{\circ}\text{C}$ ,  $t_{\text{max}}=1~\text{h}$ , air atmosphere) to accumulate chemisorption products of  $CO_2$  [6], and irradiated with gamma rays (absorbed dose 56 kGy, dose rate 14 kGy·h<sup>-1</sup>). The composition of the powder was analyzed with powder X-ray diffractometry, Fourier transformation infrared spectroscopy and thermal analysis, the accumulated radiation defects were studied with electron spin resonance spectroscopy.

It was established that X-ray amorphous chemisorption products of H<sub>2</sub>O and CO<sub>2</sub>, mostly LiOH and Li<sub>2</sub>CO<sub>3</sub>, accumulate on the surface of Li<sub>4</sub>SiO<sub>4</sub> powder after thermal treatment in air. Radiation defects of Li<sub>2</sub>CO<sub>3</sub> and LiOH are unstable and accumulate only in low amounts. Radiolysis of chemisorption products affects the radiation stability of Li<sub>4</sub>SiO<sub>4</sub> powder and increases the concentration of radiation defects of silicates, SiO<sub>4</sub><sup>3-</sup> and SiO<sub>3</sub><sup>3-</sup>, up to 50%. On the basis of the obtained results it was concluded that a Li<sub>2</sub>CO<sub>3</sub> containing layer on the surface of Li<sub>4</sub>SiO<sub>4</sub> pebbles can reduce the radiation stability and may cause tritium accumulation. It may be favourable to change the ceramic composition and to replace Li<sub>2</sub>SiO<sub>3</sub> as secondary phase by a less reactive and radiation stable phase like Li<sub>2</sub>TiO<sub>3</sub> [7].

## References:

- [1] M. Zmitko et al. Journal of Nuclear materials 417 (2011) 678-683.
- [2] G. Kizane et al. Journal of Nuclear materials 329-333 (2004) 1287-1290.
- [3] M.H.H. Kolb et al. Journal of Nuclear materials 427 (2012) 126-132.
- [4] M. Kato et al. Journal of Material Science Letters 21 (2002) 485-487.
- [5] A. Zarins et al. Journal of Nuclear Materials 429 (2012) 34-39.
- [6] J. Ostiz-Landeros et al. Thermochimica Acta 515 (2011) 73-78.
- [7] R. Knitter et al. Journal of Nuclear materials, http://dx.doi.org/10.1016/j.jnucmat.2012.10.034.